FLAVONES FROM CITRUS SUDACHI

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Previously we reported the isolation of three flavonoids, sudachitin (1)[1,2], demethoxysudachitin (2)[3] and sudachiin A (3)[4] from green peel of *Citrus sudachi*. We now wish to report the isolation of four further flavones from the ether extract of the green peel. Three of the flavones not previously reported from *C. sudachi* are dinatin (4)[5,6], xanthomicrol (5)[7] and 5,7,4'-trihydroxy-6,3'-dimethoxyflavone (6)[8,9]. The fourth, a new flavone namely 7-methylsudachitin, has been assigned the structure 5,4'-dihydroxy-6,7,8,3'-tetramethoxyflavone (7) based on the following spectral evidence.

1 $R^1 = H$, $R^2 = R^3 = OMe$

2 $R^1 = R^3 = H, R^2 = OMe$

3 $R^1 = Glu, R^2 = R^3 = OMe$

4 $R^1 = R^2 = R^3 = H$

5 $R^1 = Me, R^2 = OMe, R^3 = H$

6 $R^1 = R^2 = H, R^3 = OMe$

7 $R^1 = Me, R^2 = R^3 = OMe$

7-Methylsudachitin (7) formed yellow needles from EtOH, mp 160-61°, $C_{19}H_{18}O_8$ (M⁺, 374). Spectral data of 7 revealed the presence of a flavone skeleton with 2 OH and 4 OMe groups. Its ¹H NMR spectrum indicated that the 3-, 2'-, 5'- and 6'-positions in the flavone skeleton are unsubstituted. In the UV spectrum of 7, the band I (349 nm) in EtOH undergoes a bathochromic shift upon the addition of AlCl₃ or NaOAc. It shows that one hydroxyl group is located at the 5-position and the remaining hydroxyl group at the 7- or 4'-position. The presence of a 4'-hydroxyl group is indicated by a large bathochromic shift (420 nm) of band I and its intensity markedly greater than that of band II in the presence of NaOAc [10]. Therefore, 7 has a 5,4'dihydroxyflavone skeleton with four OMe groups. The confirmation of the structure of 7 was made by a direct comparison with an authentic sample of 5,4'-dihydroxy-6,7,8,3'-tetramethoxyflavone [11, 12].

This is the first isolation from a natural source of 7. The two flavones, 7 and 5, are the 7-methyl ethers of sudachitin (1) and demethoxysudachitin (2), respectively.

EXPERIMENTAL

¹H NMR spectrum was taken at 90 MHz in CDCl₃ and chemical shifts are given in δ (ppm) scale relative to TMS; UV spectra were obtained in EtOH.

Isolation. The fresh green peels (5 kg) of C. sudachi Hort. ex Shirai collected in Tokushima prefecture were extracted with EtOH. The EtOH concentrate was washed with hexane and then extracted with Et₂O. The extract dissolved in a small amount of MeOH, was allowed to stand in refrigerator to give a ppt., which was separated by filtration. The ppt. contained sudachitin (1) and demethoxysudachitin (2), by recrystallization. From the mother liquor, compound A (5 mg) was isolated by HPLC with the column packed with Hitachi gel No. 3011 using aq. MeOH. The MeOH filtrate was subjected to column chromatography over a polyamide using MeOH as the eluent to give three fractions. Fraction 1 was rechromatographed with Si gel (CHCl₃-EtOAc, 20:3) to give compounds B (20 mg) and C (20 mg). Fraction 3 was extracted with aq. Na₂CO₃ soln and then the extract was acidified with dilute HCl. The separated ppts were rechromatographed with a Si gel (CHCl₃-EtMeCO, 10:1) to give compound D (16 mg).

Identification. Compound B crystallized from EtOH as yellow needles, mp 160–161°: MS: M⁺ 374 (C₁₉H₁₈O₈). It gave positive FeCl₃ and Mg–HCl tests. UV $\lambda_{max}^{\text{EnOH}}$ nm (log ε): 256 (4.17); 280 (4.23), 349 (4.36); + AlCl₃ 261 (4.12), 288 (4.21), 302sh (4.19), 363 (4.39); + NaOAc 267 (4.26), 420 (4.47). ¹H NMR (CDCl₃): δ 3.94 (3 H, s, OMe), 3.97 (6 H, s, OMe), 4.10 (3 H, s, OMe), 6.57 (1 H, s, H₃), 7.02 (1 H, d, J = 9 Hz, Ar–H₅·), 7.40 (1 H, d, J = 3 Hz, Ar–H₂·), 7.54 (1 H, q, J = 9, 3 Hz, Ar–H₆·). It was identified as 5.4′-dihydroxy-6,7.8,3′-tetramethoxyflavone (7) by mp, UV, IR and direct comparison with an authentic sample. Compounds A, C and D were identified as dinatin (4) (mp 297–299°). xanthomicrol (5) (mp 225–226°) and 5,7,4′-trihydroxy-6,3′-dimethoxyflavone (6) (mp 226–227°) by direct comparison with authentic samples.

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338

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